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A new highly efficient synthetic route to enantiopure 10-bromocamphor

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Abstract

A new enantiospecific synthetic route to the interesting chiral synthetic intermediate 10-bromocamphor starting from readily available camphor is described. The procedure takes place straightforwardly in only three synthetic steps with high overall yield. Mechanistically, two interesting enantiospecific Wagner–Meerwein rearrangements involving 2-norbornyl carbocations take place during the process. © 2000 Elsevier Science Ltd. All rights reserved.

Enantiopure C10-substituted camphor derivatives have been widely used as chiral starting materials in asymmetric synthesis,¹ as well as interesting chiral synthetic intermediates in the preparation of natural products.² Unfortunately, most of these derivatives have, as a common chemical characteristic, the presence of a sulfur atom attached to the C10 position (C10–S).¹ This common presence of sulfur at C10 is due to the fact that most of the synthetic routes for the preparation of C10-substituted camphors start from 10-camphorsulfonic acid (the first C10-substituted camphor derivative obtained),³ or the now commercially available 10-camphorsulfonyl chloride.^{1–3}

Nevertheless, nowadays, several enantiopure camphor derivatives having a C10-substitution different than the common C10–S, such as C10–O (e.g. 1), C10–N (e.g. 2), C10–halogen (e.g. 3), C10–P (e.g. 4) or C10–Se (e.g. 5) (Fig. 1), have also been described as very interesting sources of chirality.⁴ In this sense, 10-bromocamphor 3 is a very convenient and a key synthetic precursor to C10-substituted camphor derivatives, since the bromine atom can be replaced via a nucleophilic substitution reaction (e.g. C10–Br→C10–O,^{4a} C10–Br→C10–Se,⁴ⁱ or C10–Br→C10–P^{4j}). Unfortunately, up to now 10-bromocamphor 3 has only been prepared enantiospecifically from 10-camphorsulfonic acid in a very low overall yield (20%), following a procedure described by

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Figure 1. Some selected C10-substituted camphors with a C10-substitution different than C10-S

Dallacker et al. in 1961.^{4a} This renders synthetic routes using 10-bromocamphor as starting material unattractive.^{4a,i,j}

In this communication, we describe a new highly efficient synthetic route to (1R)-10-bromocamphor **3** starting from natural (1R)-camphor **6**. Only three simple synthetic steps, which take place under mild reaction conditions and with excellent yields, are required (Scheme 1).

The first step is the preparation of (1R)-3,3-dimethyl-2-methylenenorborn-1-yl triflate 7 from commercial (1R)-camphor 6 by reaction with triflic anhydride.⁵ Triflate 7 is subsequently reduced with lithium aluminum hydride to give the corresponding 2-methylenenorbornan-1-ol 8.⁶ Finally, the treatment of alcohol 8 with *N*-bromosuccimimide (NBS) yields the desired (1R)-10-bromo-camphor 3.⁷ Obtaining synthetic intermediate 8 is necessary since the 2-methylenenorborn-1-yl triflate 7 does not react with NBS under the same conditions as alcohol 8.



Scheme 1. New highly efficient enantiospecific preparation of 10-bromocamphor 3

The key steps of the above-described synthetic procedure are: (1) the enantiospecific preparation of triflate 7 by Wagner–Meerwein rearrangement of 6 by reaction with triflic anhydride; and (2) a second enantiospecific Wagner–Meerwein rearrangement of the 2-methylenenorbornan-1-ol 8 by carbon–carbon double bond addition of electrophilic bromine (Br⁺), under a straightforward NBS treatment (Scheme 2).



Scheme 2. Wagner-Meerwein rearrangements involved in the described preparation of 10-bromocamphor 3

In conclusion, a new enantiospecific route to the interesting chiral source 10-bromocamphor **3** has been described. 10-Bromocamphor **3** is now obtained in only three steps starting from commercially available (1R)-camphor **6**, the complete process having an overall yield of 87%. This straightforward preparation of **3** opens the way to the efficient preparation of other important enantiopure C10-substituted camphor derivatives.

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- 7. A solution of alcohol 8 and N-bromosuccinimide in dry CH₂Cl₂ is stirred at room temperature for 24 h. After the usual work-up, 3 is obtained as a white solid (95% yield). Pf. 69–70°C. [α]_D²⁰ +14.6 (0.20, CH₂Cl₂). The structure was confirmed by ¹H NMR, ¹³C NMR, IR, MS and HRMS.